

UNIVERSITY OF CALCUTTA

SARSUNA COLLEGE

CU CBCS SEM-III CEMG PRACTICAL EXAMINATION,2020

PAPER-CC3/GE3

FULL MARKS-30

TIME DURATION-1 Hr 30Mins.

SINGLE PDF FORMAT Of the Answer Scripts Should Be Sent To The Following  
EMAIL ID

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1) Answer the following questions. 15x1=15

a)  $\text{NH}_4^+$  radical gives a brown precipitate when treated with -----  
reagent in NaOH medium.

b) The colour of the Borax bead produced with  $\text{Co}^{2+}$  salts is -----.

c) All sodium salts are water soluble---TRUE/FALSE(Tick the correct option)

d) Sulphide salts produce ----- colouration when their sodium carbonate  
extract solution was treated with sodium nitroprusside solution.

e)  $\text{Ca}^{2+}$  salts produce ----- colour in flame test.

f) In Group-IV basic radical analysis  $\text{Ba}^{2+}$  radical is precipitated as barium  
chromate( a yellow precipitate) by adding ----- solution .

g) During the flame test performing with an inorganic salt sample ,the  
platinum wire was frequently cleaned by using ---- ----- acid.

h) Ammonium ( $\text{NH}_4^+$ ) salts produce a ----- colour sublimate on the  
cooler part of the test tube during dry test tube heating test.

i) In the oxidative fusion test for  $\text{Cr}^{3+}$  and  $\text{Mn}^{2+}$  the reactant mixture was  
taken on a ----- foil.

j) The flame colour of copper salt is apple green/bluish green(tick the correct one)

k) Soda- Lime test is performed to detect the presence of----- salts.

l) Name group reagents for Group-III A.

m) A sample salt when treated with concentrated sulphuric acid( $H_2SO_4$ ) produces violet vapour, this indicates the presence of which of the following radicals :  $I^-$  OR  $Br^-$

n) Name an interfering acid radical.(with formula/symbol)

o) Sample+ dilute sulphuric acid +few drops of freshly prepared ferric chloride(  $FeCl_3$ ) solution produces blood red colouration, this observation indicates the presence of ----- radical(name the radical with formula)

2) Draw three columns with headings Experiment, Observation, Inference(in the given order) and write the following tests in detail.(an example is provided below) ( 2+2+1=5)x3=15

Experiment	Observation	Inference
Describe in detail	Describe in detail	X radical(Its name and symbol, eg. $Na^+$ , Sodium Salt) present and confirmed

a) Chromyl Chloride Test For Chloride Radical

b) Special Test For Nickel Radical with the HCl-Extract of The Sample

c) Oxidative Fusion Test For Chromium Radical